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PRESERVATION OF ORGANICS AT MARS' NEAR-SURFACE

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CONTEXT AND INTRODUCTION

One of the biggest concerns for the *in situ* detection of organic molecules on extraterrestrial environment is the preservation potential of the molecules at the surface and subsurface given the harsh radiative environment and oxidants they are exposed to.

The Mars Science Laboratory (MSL) mission hosts Sample Analysis at Mars (SAM), a 40 kg suite of instruments which is devoted to make the inventory of organic and inorganic compounds in Mars' atmosphere and subsurface, and understanding their processes of preservation.

To date, SAM has detected and identified several organic molecules in the Martian subsurface, such as chloroalkanes, chlorobenzene at various states of chlorination, sulphur-containing molecules and functionalized aromatic hydrocarbons.

The presence of organic molecules opens up habitability to another level, where the building blocks of life were available. Understanding their windows of preservation of organics will help in the search for prebiotic or biological signature on Mars.

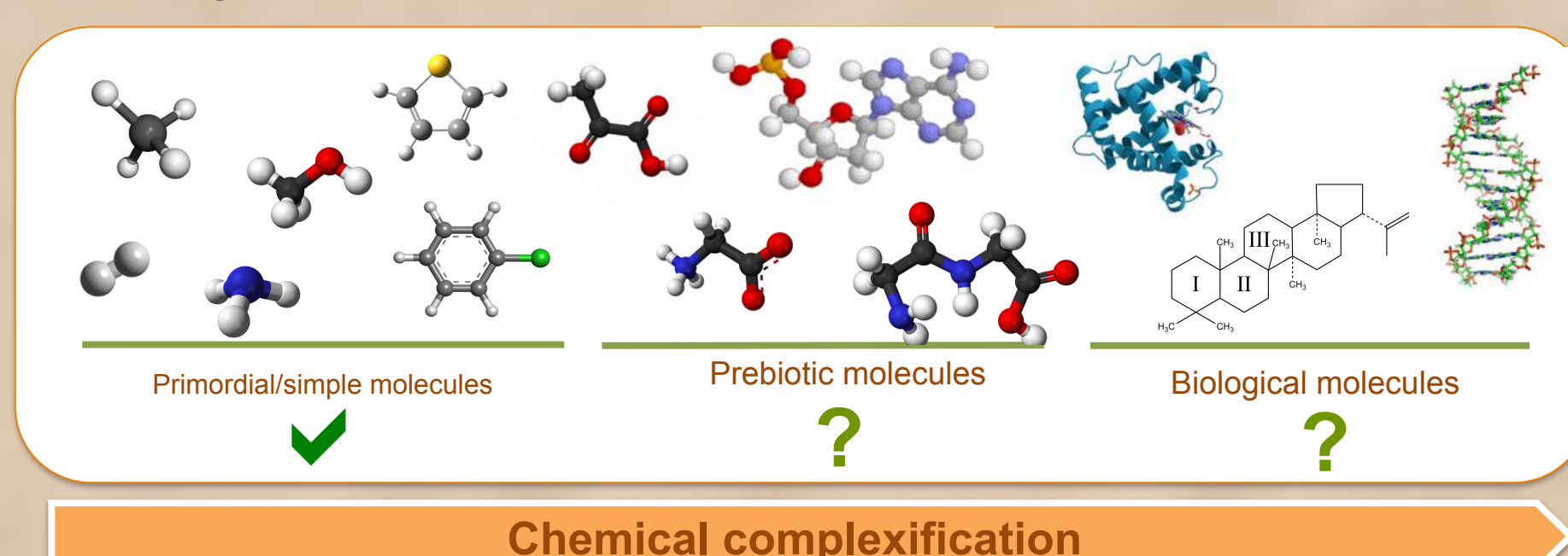


Fig. 1: From chemistry to biology, one can see a complexification of organic molecules. Looking for the whole range will assess the past and present biological potential of Mars.

MATERIAL AND METHODS

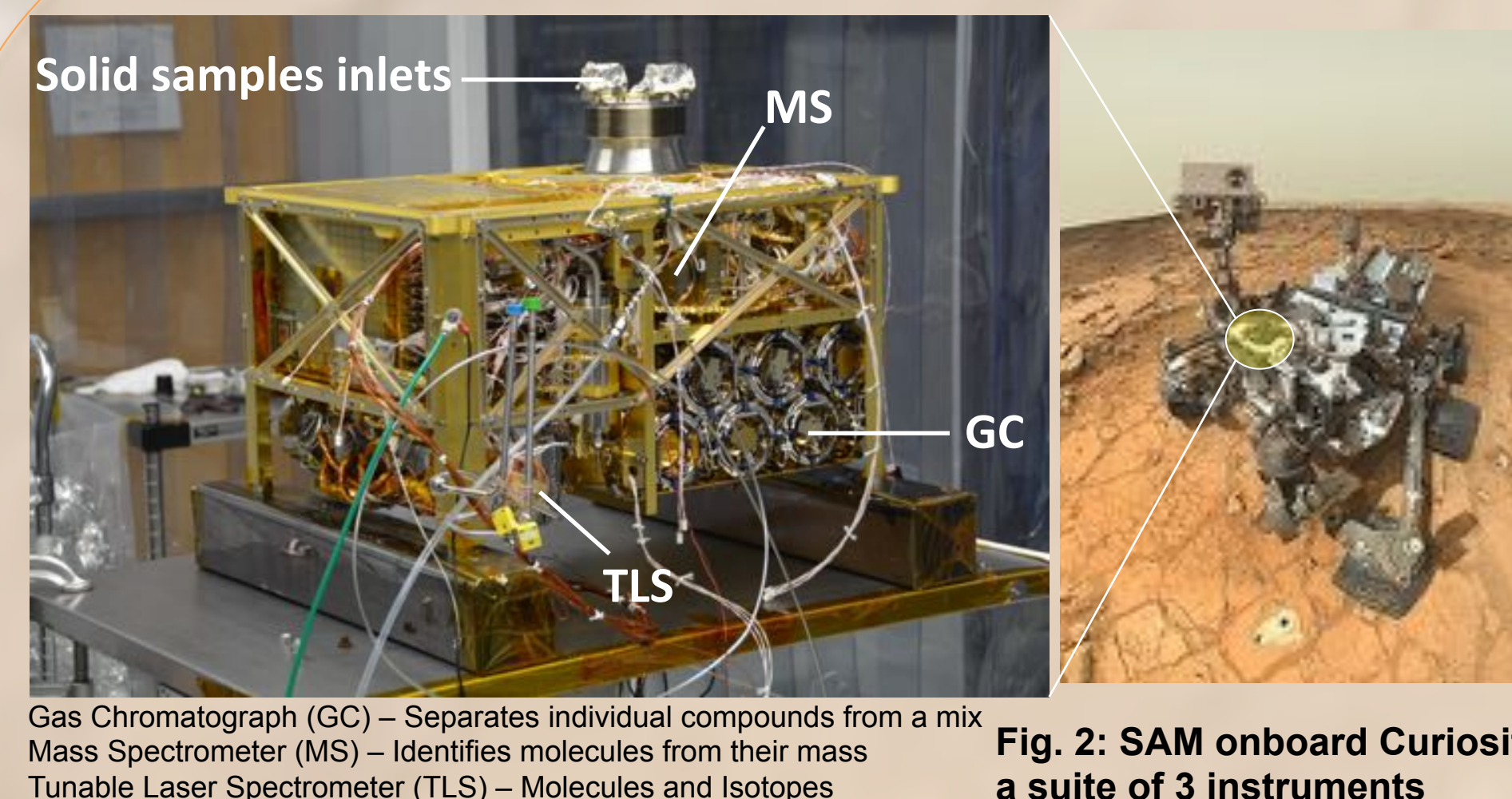


Fig. 2: SAM onboard Curiosity: a suite of 3 instruments

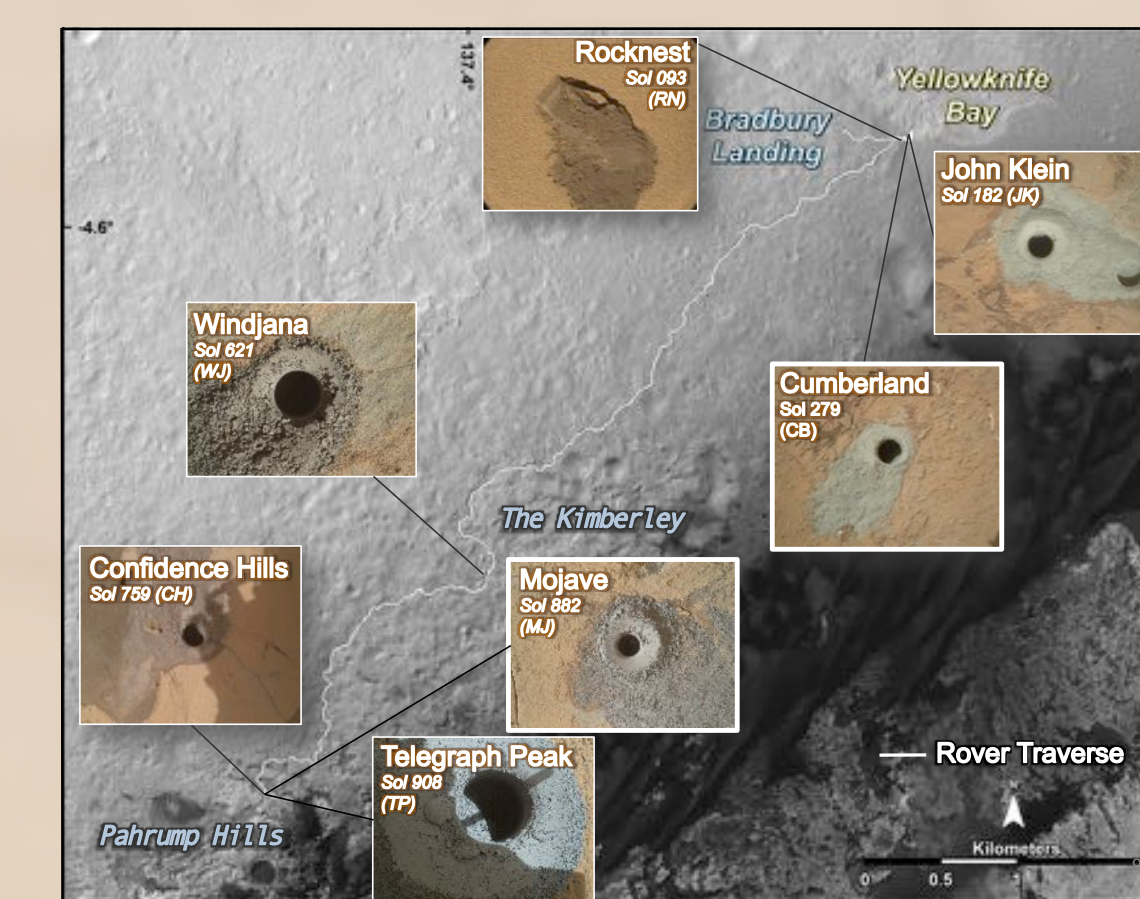


Fig. 3: MSL traverse map and drilling sites

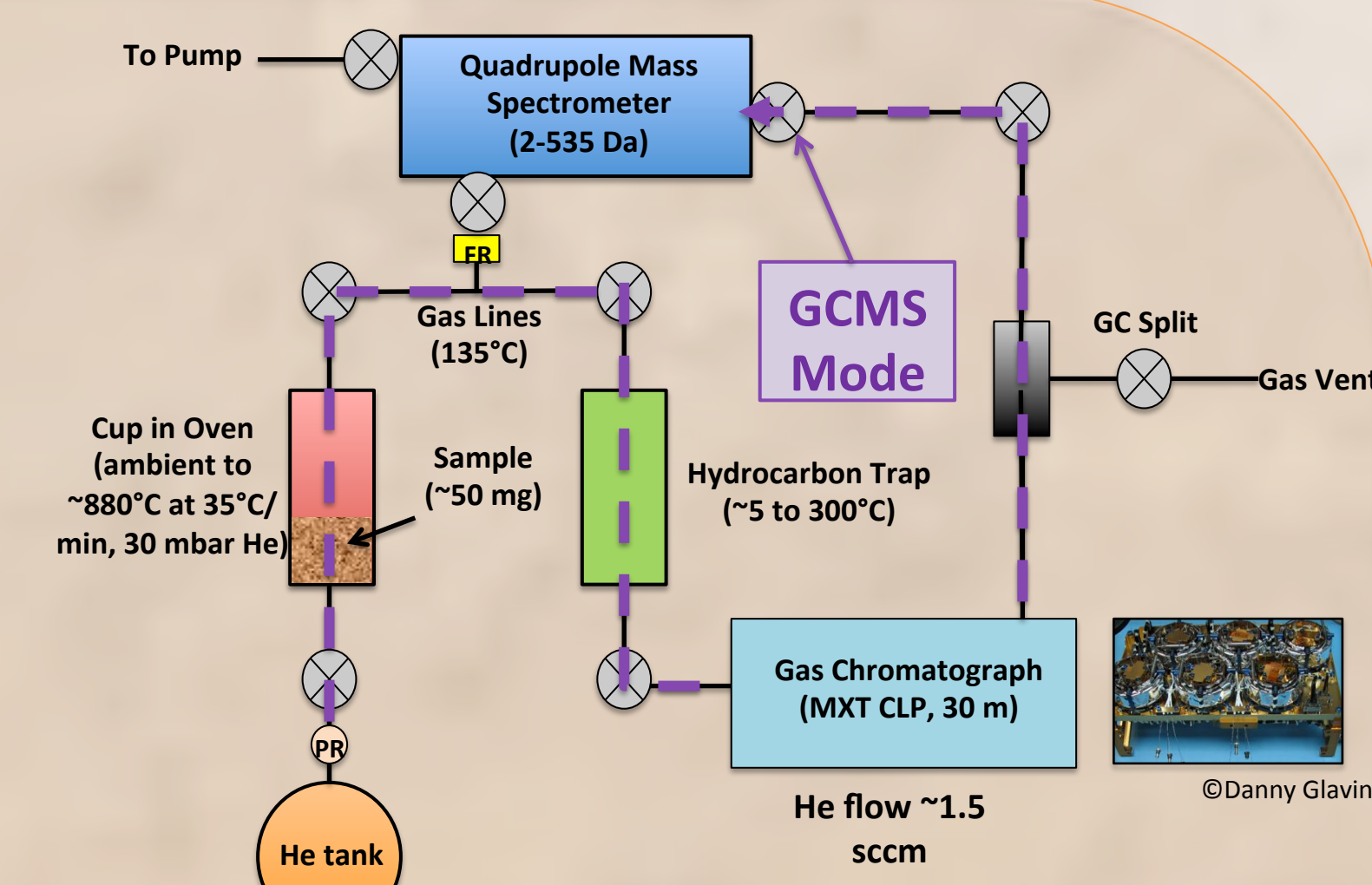


Fig. 4: Simplified gas flow diagram for the GCMS mode of SAM

SAM (Fig. 2) is a suite of three instruments that measure volatiles extracted from solid samples using three primary analytical modes for analysis of organic compounds, including: (1) Evolved gas analysis-quadrupole mass spectrometry (EGA), where solid samples are heated inside a pyrolysis oven up to ~1000°C and the gases released are monitored by quadrupole mass spectrometry (QMS), with portions isolated for gas chromatography mass spectrometry (GCMS) (Fig. 4) and (2) wet chemistry, which consists of an extraction and derivatization with N-methyl-N-tertbutyldimethylsilyl-trifluoroacetamide (MTBSTFA) followed by GCMS analysis (Fig. 5).

To date, MSL travelled almost 13 km at the surface of Mars and drilled 10 rocks for 9 SAM analyzes. The two samples discussed in this poster are Cumberland (CB), in the Sheepbed mudstone, and Mojave (MJ), Fine-grained sedimentary deposits in Murray Formation of the Pahrump Hills (Fig. 3).

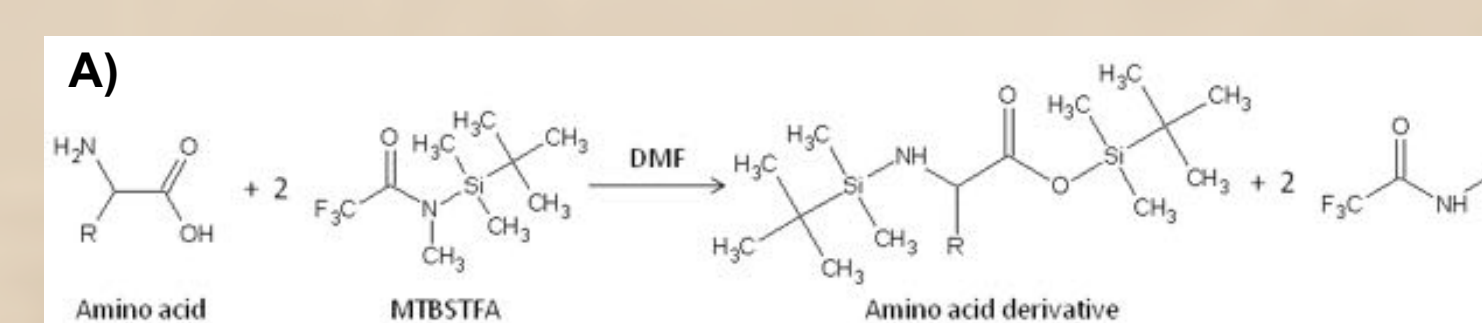
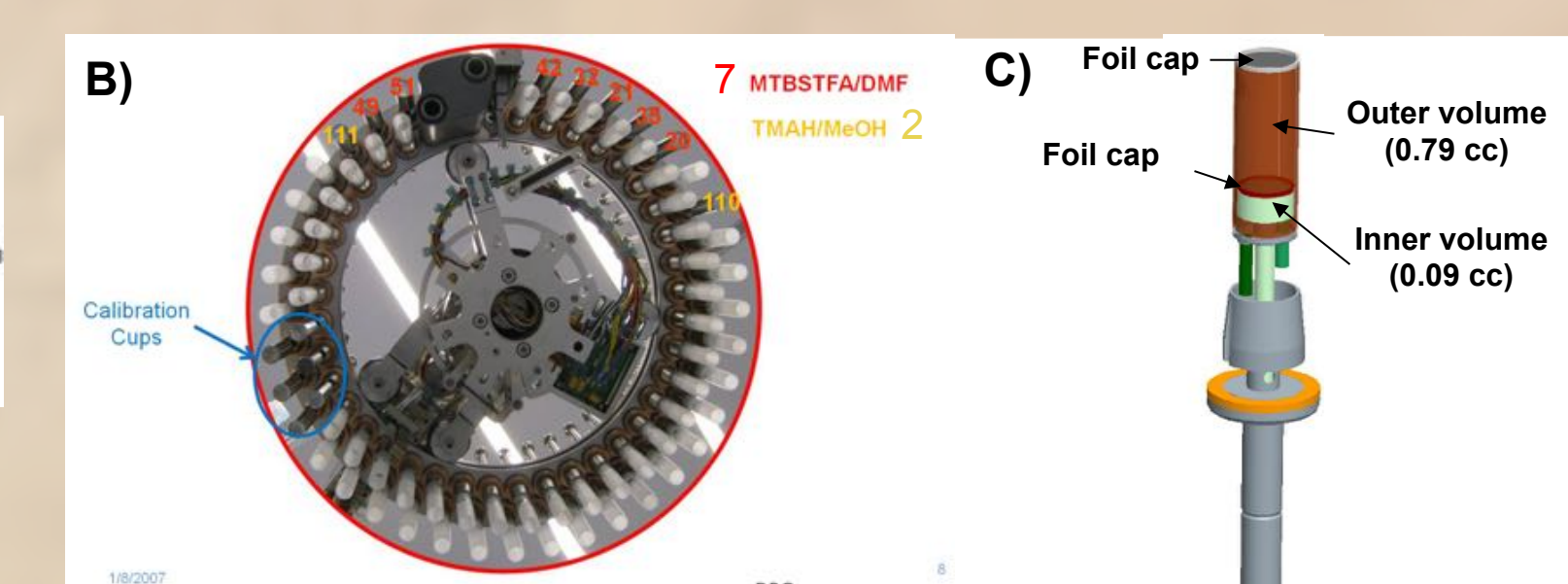


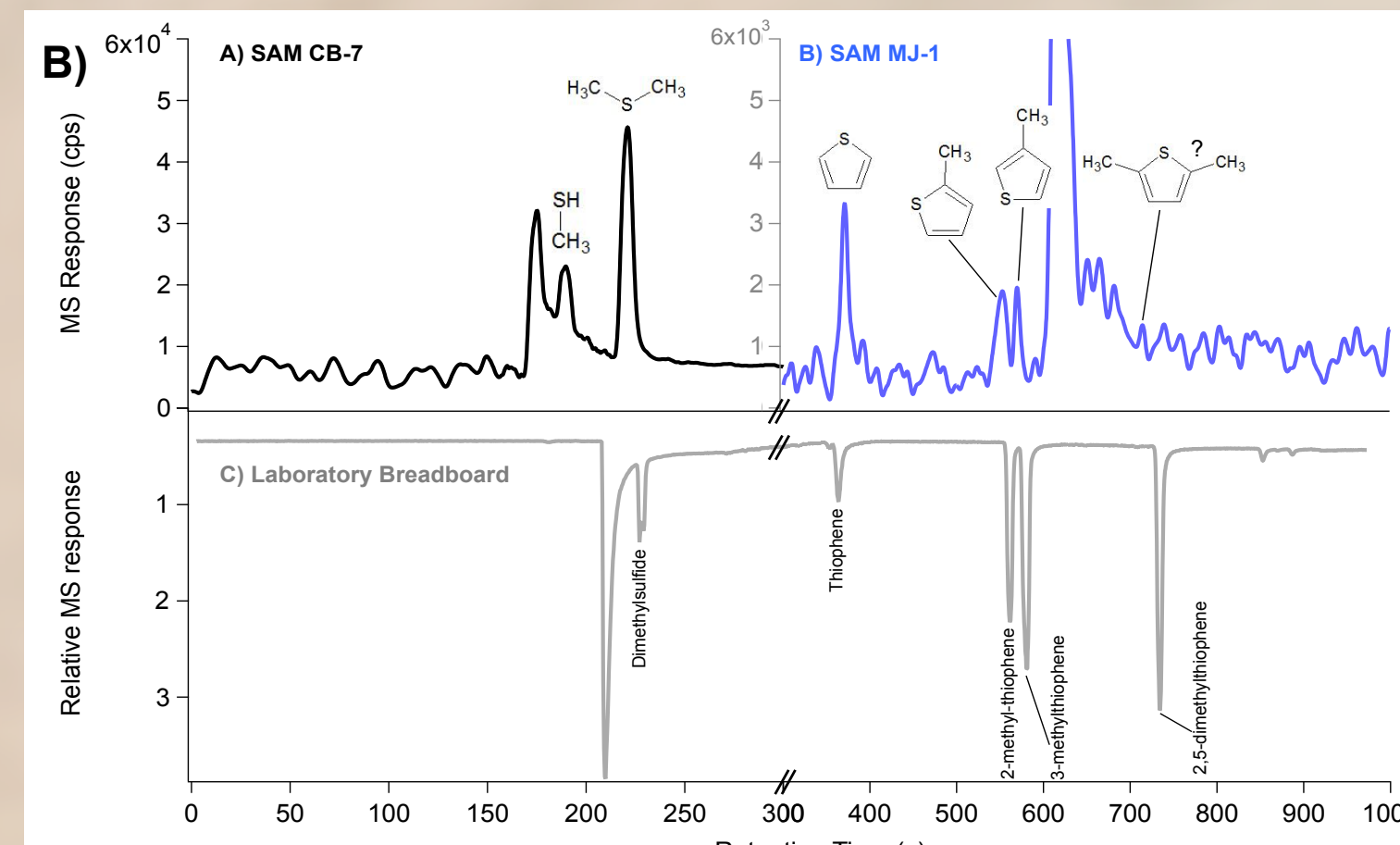
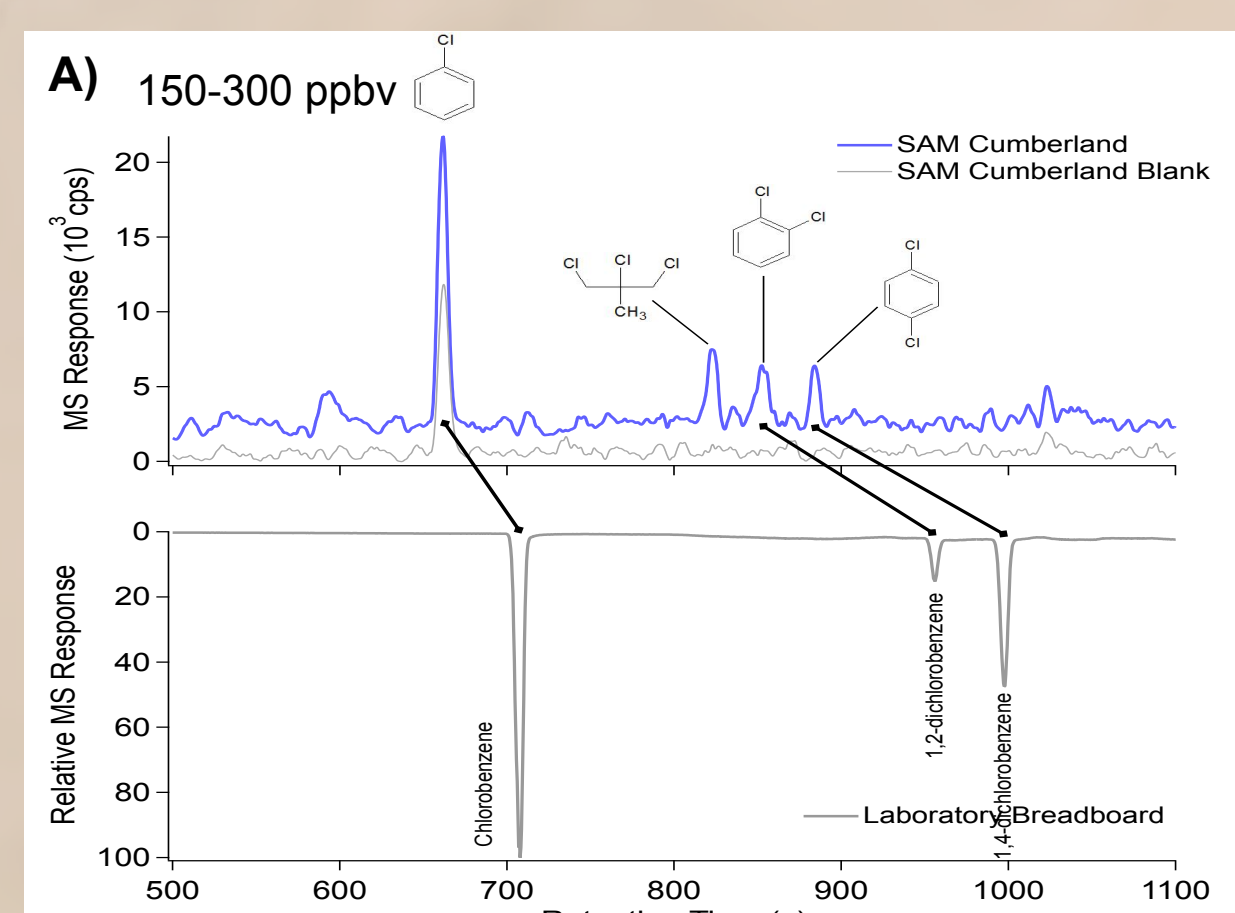
Fig. 5: SAM wet chemistry. A) MTBSTFA derivatization reaction. B) SAM carousel containing 7 MTBSTFA cups (detailed in C))



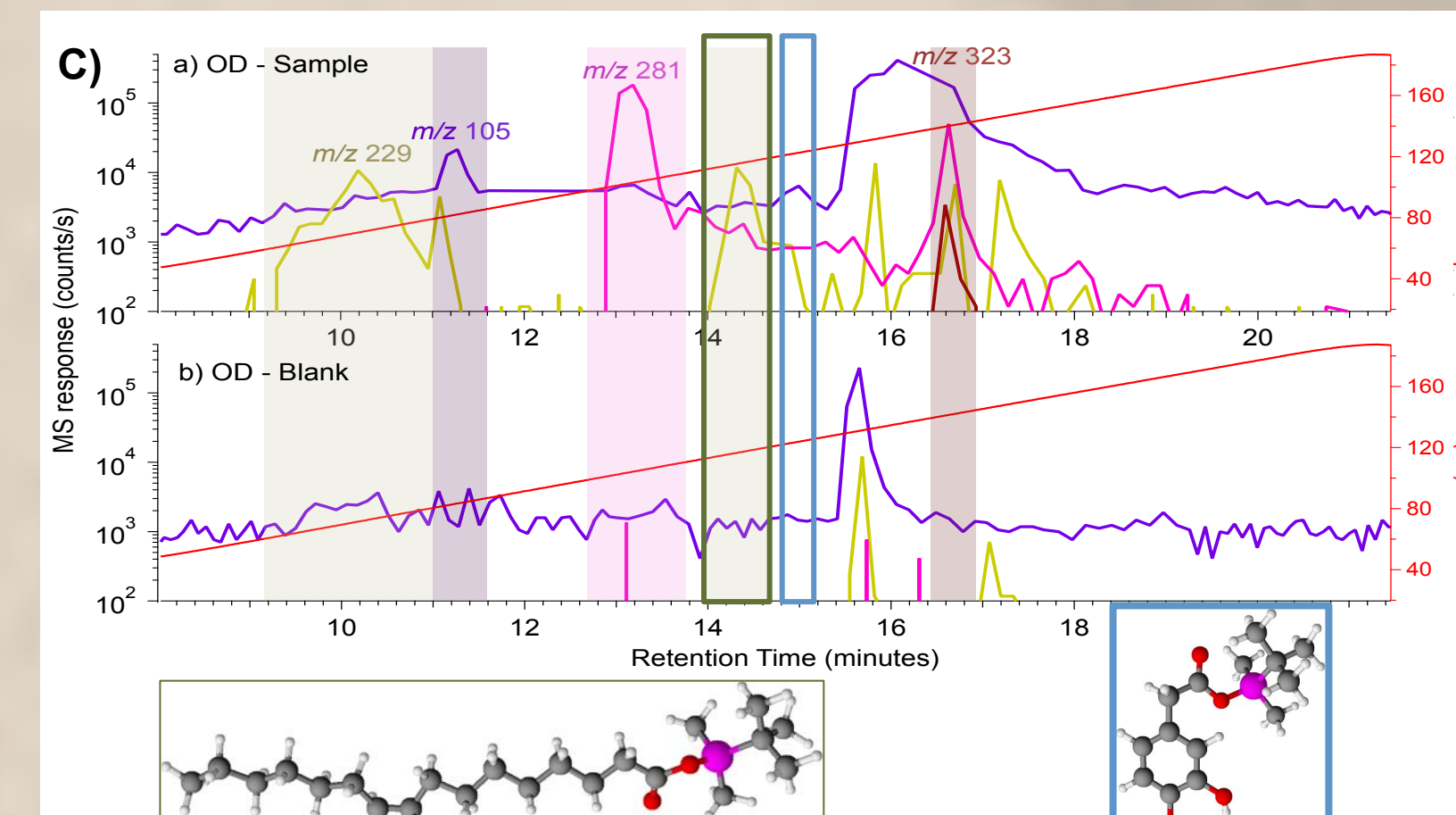
RESULTS AND DISCUSSION

1) Detection of organic molecules indigenous to Mars' subsurface

Pyrolysis-GCMS



Derivatization-GCMS



Low temperature cut

Fig. 6: SAM GCMS results. A) Low temperature cut on CB sample (< 450 °C), identification of chlorohydrocarbons from pyrolysis. B) High temperature cut on CB or MJ samples (> 600 °C), detection of sulfur-containing, non-chlorinated organics from pyrolysis. C) Derivatization GCMS on high temperature cut of CB sample (> 250 °C), detection of higher complexity, non-chlorinated organics with tentative identification

Several organics were identified on Mars, however, the origin (biological vs. abiotic) and the nature of the precursors of the chlorinated organics are unknown (6A).

Sulphur-containing compounds identified at high-temperature in CB and MJ (e.g. thiophene – 6B). Low-temp organics are chlorinated, high-temp organics are not! High-temp. organics are protected from the O₂/Cl released from the decomposition of perchlorates.

2) Origin of the chlorohydrocarbons – effect of perchlorates

The chlorohydrocarbons are expected to be formed from the reaction of a non-chlorinated precursor present in the sample and Cl from perchlorate decomposition, in SAM oven even if there presence as such in the sample cannot be totally excluded.

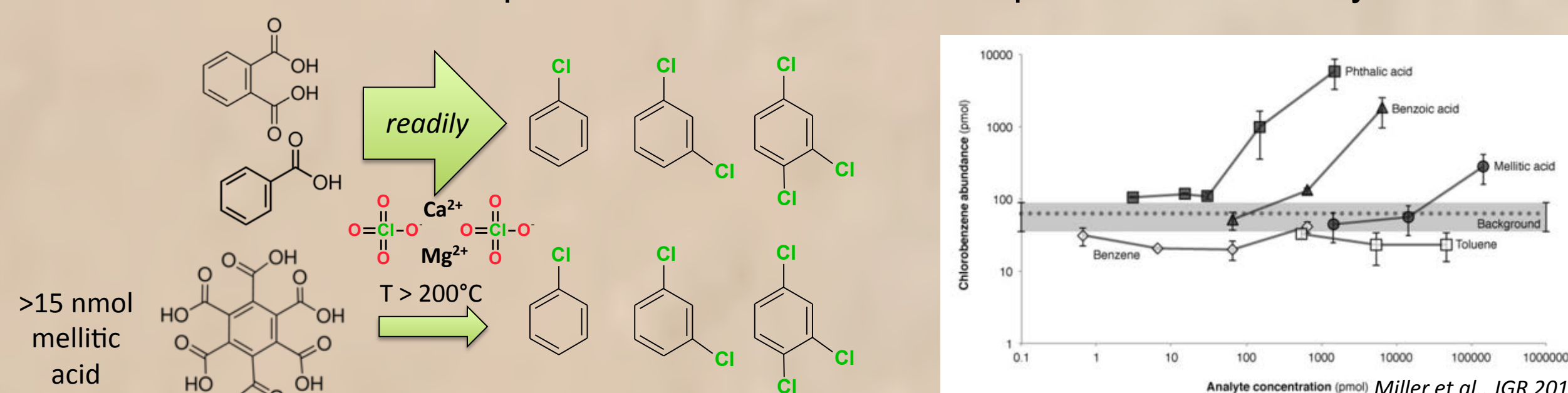


Fig. 7: laboratory experiment show that benzoic, phthalic and mellitic acid, heated in the presence of perchlorate, would form chlorobenzene, dichlorobenzene and trichlorobenzene, and that the abundance of CBZ depends on the initial abundance of the aromatic hydrocarbon.

